



Amino acid salt solutions as solvents in CO₂ capture from flue gas CO₂ loading capacity and precipitation.

Lerche, Benedicte Mai; Thomsen, Kaj; Stenby, Erling Halfdan

Publication date:
2011

Document Version
Publisher's PDF, also known as Version of record

[Link back to DTU Orbit](#)

Citation (APA):
Lerche, B. M., Thomsen, K., & Stenby, E. H. (2011). *Amino acid salt solutions as solvents in CO₂ capture from flue gas: CO₂ loading capacity and precipitation..* Poster session presented at 6th Trondheim CO₂ capture, transport and storage conference, Trondheim, Norway.

General rights

Copyright and moral rights for the publications made accessible in the public portal are retained by the authors and/or other copyright owners and it is a condition of accessing publications that users recognise and abide by the legal requirements associated with these rights.

- Users may download and print one copy of any publication from the public portal for the purpose of private study or research.
- You may not further distribute the material or use it for any profit-making activity or commercial gain
- You may freely distribute the URL identifying the publication in the public portal

If you believe that this document breaches copyright please contact us providing details, and we will remove access to the work immediately and investigate your claim.

Amino acid salt solutions as solvents in CO₂ capture from flue gas; CO₂ loading capacity and precipitation.

Benedicte Mai Lerche^a, Erling H. Stenby^b and Kaj Thomsen^a

CERE (Center for Energy Resources Engineering) ^aDepartment of Chemical and Biochemical Engineering. ^bDepartment of Chemistry, DTU (Technical University of Denmark) Søltofts plads building 229, 2800 Kgs. Lyngby. DK

E-mail: bml@kt.dtu.dk

Keywords : CO₂ capture; amino acid salt solutions; CO₂ loading capacity; Precipitation

New solvents based on the salts of amino acids have emerged as an alternative to the alkanolamine solutions, for the chemical absorption of CO₂ from flue gas. But only few studies on amino acids as CO₂ capturing agents have been performed so far. One of the interesting features of amino acid salt solutions is their ability to form solid precipitates upon the absorption of CO₂. The occurrence of crystallization offers the possibility of increasing the CO₂ loading capacity of the solvent. However, precipitation can also have negative effect on the CO₂ capture process. The chemical nature of the solid formed is a decisive factor in determining the effect of precipitation on the process.

For the purpose of studying the CO₂ loading capacity of amino acid salt solutions, we developed an experimental set-up based on a dynamic analytical mode, with analysis of the effluent gas. Using this set-up, the CO₂ loading capacity of aqueous solutions of the potassium salts of selected amino-acids (glycine, taurine, lysine proline, and glutamic acid) were examined, and the relation between the initial amino acid salt concentration and precipitation ability of each solution were determined. Experiments were performed at a partial pressure of CO₂ close to 10 kPa, and a total pressure around 100 kPa, and a temperature close to 298 K. The obtained precipitates were analyzed using X-ray diffraction and infra-red spectroscopy. It was verified that the precipitate consisted of the amino acid itself in the case of glycine, taurine, and lysine, while in the case of proline, and glutamic acid, the precipitate was found to be bicarbonate. These results give an important contribution to further understanding the potential of amino acid salt solutions in CO₂ capture from flue gas.